BBA Report

BBA 71104

Ca2+-activated membrane ATPase: Selective inhibition by ruthenium red

EILEEN L. WATSON, FRANK F. VINCENZI and PAUL W. DAVIS

Department of Pharmacology, School of Medicine and College of Pharmacy, University of Washington, Seattle, Wash. $98105\ (U.S.A.)$

(Received September 20th, 1971)

SUMMARY

- 1. Ca²⁺-ATPase, (Na⁺-K⁺)-ATPase and Mg²⁺-ATPase activities were determined in isolated red blood cell membranes.
- 2. The hexavalent dye, ruthenium red, produced concentration-dependent inhibition of Ca^{2+} -ATPase ($I_{50} \cong 10^{-5}$ M) without significantly affecting (Na⁺-K⁺)-ATPase or Mg²⁺-ATPase.
- 3. We suggest that ruthenium red may be a useful tool for selective inhibition of Ca²⁺ transport and/or Ca²⁺-ATPase.

Since the report by Schatzmann¹ concerning active Ca²⁺ transport in red cell ghosts, there has been increasing evidence for such transport^{2,3} and its relationship to a membrane-bound enzyme, i.e. a Ca²⁺-activated, Mg²⁺-dependent ATPase (Ca²⁺-ATPase)^{4,5}. Ca²⁺-ATPase of erythrocyte membranes is not inhibited by ouabain⁵⁻⁷ which inhibits Na⁺,K⁺-activated, Mg²⁺-dependent ATPase ((Na⁺-K⁺)-ATPase)^{6,8} and Na⁺,K⁺ transport in the red cell⁸⁻¹⁰. Oligomycin which also inhibits (Na⁺-K⁺)-ATPase⁶ and Na⁺ transport¹¹ is ineffective as an inhibitor of calcium transport in erythrocytes^{2,5}. The search for a specific inhibitor of Ca²⁺-ATPase and/or Ca²⁺ transport has thus far been unsuccessful. Mersalyl⁵, ethacrynic acid¹², and the lanthanides¹³ inhibit Ca²⁺-ATPase and Ca²⁺ transport. However, these substances are rather non-specific in that they inhibit other membrane ATPases such as (Na⁺-K⁺)-ATPase^{5,13}.

Recently, ruthenium red, an inorganic dye used in the staining of tissues, cells and organelles¹⁴⁻¹⁶, has been shown to inhibit Ca²⁺ transport in isolated mitochondria¹⁷. Thus, in searching for a possible inhibitor of Ca²⁺-ATPase or Ca²⁺ transport, we decided to examine the effects of ruthenium red on Ca²⁺-ATPase, (Na⁺-K⁺)-ATPase and Mg²⁺-dependent ATPase (Mg²⁺-ATPase) in erythrocyte membranes.

Abbreviation: EGTA, ethyleneglycol-bis-(\beta-aminoethyl ether) N,N'-tetraacetic acid.

Biochim. Biophys. Acta, 249 (1971) 606-610

Isolated human red cell membranes were prepared as described previously^{18,19} with the slight modification that outdated (22-28 day old) packed red cells were used instead of whole blood.

ATPase activities were determined as follows; 0.3 ml of red cell membrane suspension was preincubated with 80 mM NaCl, 15 mM KCl, 3 mM MgCl₂, 4 mM ethyleneglycol-bis-(β -aminoethyl ether) N, N'-tetraacetic acid (EGTA), 30 mM Tris-HCl and ruthenium red at pH 7.1 in a final volume of 2 ml at 37° for 5 min. Choline chloride (95 mM) was substituted for NaCl and KCl in some experiments. Tris-ATP (3 mM) was added to start the reaction and the mixture was incubated at 37° for 60 min. Mg²⁺-ATPase was determined with the addition of 0.1 mM ouabain to the incubation medium. (Na⁺-K⁺)-ATPase, *i.e.* the ouabain-sensitive ATPase, was calculated as the difference between activities in the presence and absence of ouabain. Ca²⁺-ATPase, known to be insensitive to ouabain⁵, was calculated as the difference produced by the addition of Ca²⁺ (pCa²⁺ = 3.92) in the presence of ouabain. EGTA (4 mM) was present in all incubations including the Ca²⁺-free conditions. Standard Ca²⁺ stock solutions were prepared by compleximetric titration. Aqueous solutions of ruthenium red (K and K Laboratories) were prepared and assayed for purity and concentration according to the method of Luft²⁰.

The enzyme reaction was terminated by the addition of 1.0 ml of 1.5 M HClO₄. After removal of the precipitated protein by centrifugation, an aliquot of the incubation medium was assayed for inorganic phosphate by the method of Fiske and SubbaRow²¹ with correction by means of appropriate blanks for non-enzymatic hydrolysis. Ruthenium red did not interfere with the assay of inorganic phosphate. Under the above conditions, ATP hydrolysis was a linear function of incubation time and enzyme concentration. Enzymatic activity was expressed as µmoles of inorganic phosphate liberated per mg protein in 60 min. The enzyme preparations were assayed for protein content according to the method of Lowry et al.²²

Control values for Ca^{2^+} -ATPase, (Na^+-K^+) -ATPase and Mg^{2^+} -ATPase were 0.90 \pm 0.19, 0.25 \pm 0.01 and 0.13 \pm 0.02 μ moles P_i per mg protein per h, respectively. Ruthenium red, a hexavalent cation, inhibited Ca^{2^+} -ATPase without significantly affecting (Na^+-K^+) -ATPase or Mg^{2^+} -ATPase (Fig. 1). The inhibitory effect on Ca^{2^+} -ATPase was observed at all concentrations of ruthenium red tested.

Several workers²³⁻²⁵ recently suggested that there are a number of Ca²⁺-ATPases present in red cell membranes. Schatzmann and Rossi²⁵ described an operational subdivision of erythrocyte membrane Ca²⁺-ATPase based on the presence or absence of Na⁺ and/or K⁺ in the incubation media. Part of the membrane Ca²⁺-ATPase is dependent on the presence of either one of the alkali cations; part of the Ca²⁺-ATPase appears to be independent of Na⁺ and/or K⁺. The functions of the various Ca²⁺-ATPases, as defined by Schatzmann, are not yet clear, although it was suggested that the Na⁺/K⁺-independent fraction is sufficient to support Ca²⁺ transport²⁵. Since in our study ruthenium red specifically inhibited Ca²⁺-ATPase rather than (Na⁺-K⁺)-ATPase or Mg²⁺-ATPase, the effects of ruthenium red were examined on each of the above described Ca²⁺-ATPase fractions. The alkali-cation-independent fraction of Ca²⁺-ATPase was measured in the manner described above for Ca²⁺-ATPase but with the use of 95 mM choline in place of omitted Na⁺ and K⁺. The alkali-cation-dependent fraction of Ca²⁺-ATPase was calculated as the difference

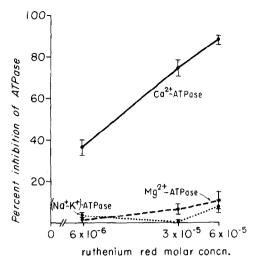


Fig. 1. ATPase activity of erythrocyte membranes in the presence of ruthenium red. Ca²⁺-ATPase, (Na⁺-K⁺)-ATPase and Mg²⁺-ATPase activities of human red cell membranes were measured as described in text. Values are means (± S.E. expressed as vertical bars) of three or five experiments.

between Ca²⁺-ATPase and alkali-cation-independent Ca²⁺-ATPase. Both types of Ca²⁺-ATPase were measured in the presence of ouabain thereby eliminating any contribution by (Na⁺-K⁺)-ATPase.

The results in Table I show that, at a concentration of $6 \cdot 10^{-6}$ M, ruthenium red inhibited each fraction by approximately one-third and was almost completely inhibitory at $6 \cdot 10^{-5}$ M. Thus, there was no apparent difference in the inhibition by ruthenium red of these two operationally defined Ca^{2+} -ATPases. Since there was no difference in the sensitivity of these ATPases to ruthenium red, they will be considered together hereafter simply as Ca^{2+} -ATPase. (Na⁺-K⁺)-ATPase and Mg²⁺-ATPase again were not affected by ruthenium red as shown earlier in Fig. 1.

TABLE I EFFECTS OF RUTHENIUM RED ON ATPases OF ERYTHROCYTE MEMBRANES

Values are expressed as μ moles of inorganic phosphate liberated per mg protein in 1 h. See text for procedural details.

ATPases	ATPase activity Ruthenium red concentration (M):		
	Ca ²⁺ -ATPase	0.88	0.55
alkali-cation-dependent	0.32	0.22	0.04
alkali-cation-independent	0.56	0.33	0.05
(Na ⁺ -K ⁺)-ATPase	0.28	0.26	0.25
(Na ⁺ -K ⁺)-ATPase Mg ²⁺ -ATPase	0.11	0.11	0.10

The mechanism of the inhibition of $\operatorname{Ca^{2^+}}$ -ATPase is not defined by our results although several possibilities can probably be ruled out. Moore 17 substituted RuCl₂ for ruthenium red and found RuCl₂ to be much less effective when compared with ruthenium red, thereby ruling out ruthenium ion (Ru³⁺) as being responsible for his results. Since ruthenium red is a hexavalent cation, chelation of $\operatorname{Ca^{2^+}}$ also appears unlikely. The presence in our system of a large reservoir of $\operatorname{Ca^{2^+}}$ chelated with EGTA also makes chelation an unlikely basis for the effects of ruthenium red which were observed in this study.

Schatzmann recently studied the effects of the lanthanides, holmium (Ho3+) and praseodymium (Pr3+) as inhibitors of Ca2+ transport in human red cell ghosts and of Ca²⁺-ATPase in isolated red cell membranes¹³. The concentrations for 50% inhibition of Ca²⁺ transport and Ca²⁺-ATPase were approximately 10⁻⁴ M. Concentrations of lanthanides that inhibited Ca²⁺-ATPase of isolated membranes also inhibited (Na⁺-K⁺)-ATPase, but did not inhibit Mg2+-ATPase. Thus, while the lanthanides were offered as a tool for inhibition of Ca²⁺ transport, they would also be expected to directly inhibit Na⁺, K⁺ transport. Our findings demonstrate that, in contrast to the lanthanides, ruthenium red inhibits Ca²⁺-ATPase without significantly inhibiting (Na⁺-K⁺)-ATPase. The concentration for 50% inhibition by ruthenium red was approximately 10⁻⁵ M. On the basis of the ATPase model for cation transport and because ruthenium red specifically inhibits erythrocyte membrane Ca2+-ATPase, we suggest that ruthenium red would inhibit Ca2+ transport without directly affecting Na+,K+ transport. Some indirect alteration of Na⁺,K⁺ transport by ruthenium red might be expected with accumulation of internal Ca²⁺ (ref. 19). It is also noteworthy that ruthenium red did not affect Mg²⁺-ATPase at concentrations which significantly inhibited Ca2+-ATPase. Duncan26 and Bowler and Duncan²⁷ have suggested that Mg²⁺-ATPase is involved in control of passive permeability of excitable cells and erythrocytes. Thus agents which, by virtue of their non-specificity, would inhibit Mg²⁺-ATPase as well as Ca²⁺-ATPase, might be expected to alter membrane permeability to various ions, including Ca²⁺. This has been noted for mersalyl⁵. Since ruthenium red did not inhibit Mg²⁺-ATPase, we have no a priori reason to expect that it would alter membrane permeability in transport studies.

The site of action of ruthenium red as an inhibitor of Ca²⁺-ATPase is not defined by these experiments. With the isolated erythrocyte membranes used in this study, both inside and outside surfaces are available to the incubation medium. It is therefore not clear with which one, or both, surfaces ruthenium red must interact to inhibit Ca²⁺-ATPase. Since ruthenium red is essentially an extracellular marker in histological work²⁰, it would not be expected to reach inner cell surfaces readily if simply added to a suspension of erythrocytes. However, if necessary in the course of transport experiments, ruthenium red could be added to the inside of erythrocytes by reversible hemolysis⁵.

We suggest that the relatively high potency and specificity of ruthenium red as an inhibitor of Ca²⁺-ATPase may allow its use as a tool for studying Ca²⁺ transport in much the same manner that ouabain is used for studying Na⁺, K⁺ transport.

We thank Dr. Ken Izutsu for alerting us to the properties of ruthenium red and for providing initial stock samples. Helpful suggestions by Dr. John Luft and equipment loan from Dr. Mont Juchau are gratefully acknowledged. This work was supported in part by a National Institutes of Health Fellowship 1 F02 AM48709-01 from the National Institute of Arthritis and Metabolic Diseases.

REFERENCES

- 1 H.J. Schatzmann, Experientia, 22 (1966) 364.
- 2 K.S. Lee and B.C. Shin, J. Gen. Physiol., 54 (1969) 713.
- 3 E.J. Olson and R.J. Cazort, J. Gen. Physiol., 53 (1969) 311.
- 4 F.F. Vincenzi and H.J. Schatzmann, Helv. Physiol, Pharmacol, Acta, 25 (1967) CR233.
- 5 H.J. Schatzmann and F.F. Vincenzi, J. Physiol. London, 201 (1969) 369.
- 6 E.T. Dunham and I.M. Glynn, J. Physiol. London, 156 (1961) 274.
- 7 P. Wins and E. Schoffeniels, Biochim. Biophys. Acta, 120 (1966) 341.
- 8 R. Whittam, K.P. Wheeler and A. Blake, Nature, 203 (1964) 720.
- 9 H.J. Schatzmann, Helv. Physiol. Pharmacol. Acta, 11 (1953) 346.
- 10 I.M. Glynn, J. Physiol., 136 (1957) 148.
- 11 I.M. Glynn, Biochem. J., 84 (1962) 75P.
- 12 F.F. Vincenzi, Proc. Western Pharmacol. Soc., 11 (1968) 58.
- 13 H.J. Schatzmann and M. Tschabold, Experientia, 27 (1971) 59.
- 14 G.T. Gustafson and E. Pihl, Acta Pathol. Microbiol. Scand., 68 (1967) 393.
- 15 J.H. Luft, J. Cell Biol., 27 (1965) 61A.
- 16 W. Bondareff, J. Neurosurg., 32 (1970) 145.
- 17 C.L. Moore, Biochem, Biophys. Res. Commun., 42 (1971) 298.
- 18 R.L. Post, C.R. Merritt, C.R. Kinsolving and C.D. Albright, J. Biol. Chem., 235 (1960) 1796.
- 19 P.W. Davis and F.F. Vincenzi, Life Sci., 10 (1971) 401.
- 20 J.H. Luft, Anat. Rec., in the press.
- 21 C.H. Fiske and Y. SubbaRow, J. Biol. Chem., 66 (1925) 375.
- 22 O.H. Lowry, N.J. Rosebrough, A.L. Farr and R.J. Randall, J. Biol. Chem., 193 (1951) 265.
- 23 V.T. Marchesi and E. Steers, Science, 159 (1968) 203.
- 24 A.S. Rosenthal, F.M. Kregenow and H.L. Moses, Biochim. Biophys. Acta, 196 (1970) 254.
- 25 H.J. Schatzmann and G.L. Rossi, Biochim. Biophys. Acta, 241 (1971) 379.
- 26 C.J. Duncan, The Molecular Properties and Evolution of Excitable Cells, Pergamon Press, 1967, p. 55.
- 27 K. Bowler and C.J. Duncan, J. Cell. Physiol., 70 (1967) 121.

Biochim. Biophys. Acta, 249 (1971) 606-610